Statistical Mechanics

August 21, 2021

Work 2 (and only 2) of the 3 problems. Please put each problem solution on a separate sheet of paper and your name on each sheet.

Problem 1

The bond distances in NH_3 are equal to 101.4 pm, and the bond angles are equal to 107.3° .

- a.) Find the location of the center of mass and the principal moments of inertia.
- b.) What is the symmetry number?
- c.) Calculate the rotational partition function of NH_3 at 500 K.

Problem 2

At low densities, the virial expansion of the equation of state of a monatomic gas can be truncated at first order,

$$PV = NkT\left(1 + \frac{N}{V}B(T) + \ldots\right)$$

with a function of temperature B(T) that you can assume to be given.

Also the heat capacity will have corrections to its ideal gas value. We can write it in the form

$$C_V = Nk\left(\frac{3}{2} - \frac{N}{V}D(T) + \ldots\right)$$

Find the form of D(T) in terms of B(T) to make these two equations thermodynamically consistent, and give also the expressions for the entropy and the internal energy to this order in terms of B(T).

Problem 3



Consider some number N of electrons with spin quantum number m_s either +1/2 or -1/2 (written as \uparrow or \downarrow), split into two different energy levels via a uniform external magnetic field **H** as shown in the above figure. This energy splitting is caused by the alignment of the electrons in either a parallel or an antiparallel configuration with respect to the magnetic field direction. Assume $E_0 = 0$.

- a.) Determine both N_{\downarrow} and N_{\uparrow} , as well as the ratio $N_{\downarrow}/N_{\uparrow}$ using the expected Boltzmann distributions at a temperature T with a magnetic field of strength H.
- b.) Determine the total magnetization M of the system as a function of N, H, and T. The magnetization is defined as $M = (N_{\downarrow} N_{\uparrow})\mu_B$. Eliminate any occurrence of N_{\downarrow} or N_{\uparrow} in your final equation.
- c.) In the limit of $H \ll T$, simplify your equation. In this limit, the equation can be expressed as $M = C\frac{H}{T}$, where C is called the *Curie constant*. Determine an equation for the Curie constant C.

Hint: The Boltzmann distribution gives $n_i = Ae^{-\epsilon_i/kT}$, where A satisfies $\sum n_i = N$